Inorg. Chem. 2007, 46, 4177-4186



Three Novel Phases in the Sm–Co–Ga System. Syntheses, Crystal and Electronic Structures, and Electrical and Magnetic Properties

YongZhong Jia, Claude Belin,* Monique Tillard,* Laurence Lacroix-Orio, David Zitoun, and GuoHua Feng

Institut Charles Gerhardt, UMR 5253 CNRS UM2, AIME, Sciences et Techniques du Languedoc, Université de Montpellier II, CC15, 2 Place Eugène Bataillon, 34095 Montpellier Cedex 5, France

Received January 16, 2007

This paper presents the synthesis, identification, and characterization of three novel phases in the ternary system Sm–Co–Ga: SmCoGa₅ (tP7, HoCoGa₅ type, tetragonal *P4/mmm*, *Z* = 1, *a* = 4.2419(3) Å, and *c* = 6.8559(5) Å), Sm₄Co₃Ga₁₆ (tP23, tetragonal *P4/mmm*, *Z* = 1, *a* = 6.0620(5) Å, and *c* = 11.1495(9) Å), and SmCoGa₄ (oC24, YNiAl₄ type, orthorhombic *Cmcm*, *Z* = 4, *a* = 4.1246(6) Å, *b* = 15.608(2) Å, and *c* = 6.4556(9) Å). The structure of SmCoGa₅ was obtained from a multiphase X-ray powder Rietveld refinement whereas the crystal structures of the other two phases were determined from single-crystal X-ray analysis. Electronic structures were calculated for all phases by first-principles DFT methods. The atomic arrangements and bonding are discussed on the basis of the partial anionic networks involving Co and Ga atoms, and a strong structural correlation is observed between SmCoGa₅ and Sm₄Co₃Ga₁₆. The latter, which displays paramagnetic behavior, has a resistivity of 4.2 $\mu\Omega$ ·cm at 3 K and undergoes a superconducting transition at 2.8 K.

Introduction

Recently, a new class of actinide-based superconductors has been discovered. The compound PuCoGa₅ displays a very high critical temperature $T_c = 18.5$ K that increases to 22 K under pressure.^{1,2} It has been described with an unconventional antiferromagnetically mediated superconductivity,³ but the superconducting mechanism is not completely elucidated and still raises debate.^{4–9} Noteworthy is the phonon-mediated

- Sarrao, J. L.; Morales, L. A.; Thompson, J. D.; Scott, B. L.; Stewart, G. R.; Wastin, F.; Rebizant, J.; Boulet, P.; Colineau, E.; Lander, G. H. *Nature* 2002, 420, 297.
- (2) Griveau, J. C.; Pfleiderer, C.; Boulet, P.; Rebizant, J.; Wastin, F. J. Magn. Magn. Mater. 2004, 272–276, 154,
- (3) Curro, N. J.; Caldwell, T.; Bauer, E. D.; Morales, L. A.; Graf, M. J.; Bang, Y.; Balatsky, A. V.; Thompson, J. D.; Sarrao, J. L. *Nature* 2005, 434, 622.
- (4) Maehira, T.; Hotta, T.; Ueda, K.; Hasegawa, A. Phys. Rev. Lett. 2003, 90 (20), 207007.
- (5) Soderlind, P. Phys. Rev B: Condens. Matter Mater. Phys. 2004, 70, 094515.
- (6) Bang, Y.; Balatsky, A. V.; Wastin, F.; Thompson, J. D. Phys. Rev. 2004, 70, 104512.
- (7) Raymond, S.; Piekarz, P.; Sanchez, J. P.; Serrano, J.; Krisch, M.; Janousova, B.; Rebizant, J.; Metoki, N.; Kanako, K.; Jochym, P. T.; Oles, A. M.; Parlinski, K. *Phys. Rev. Lett.* **2006**, *96*, 237003.
- (8) Opahle, I.; Oppeneer, P. M. Phys. Rev. Lett. 2006, 90, 157001.

10.1021/ic0700755 CCC: \$37.00 © 2007 American Chemical Society Published on Web 04/20/2007

conventional superconductivity that occurs in the same temperature range, at 18 K for Nb₃Sn and 39 K for MgB₂. Attention was also paid to other potential actinide compounds of the PuCoGa₅ family that crystallize in the HoCoGa₅-type structure.¹⁰ The physics and the chemistry of Ce-based intermetallic compounds have been extensively studied, and more than 200 papers were published last year. The CeMIn₅ (M = Co, Rh, Ir)^{11,12} family is a special class among the heavy-fermion (HF) materials that has considerably advanced the knowledge of the interplay between superconductivity and magnetism.³ These compounds, which show magnetic ordering and unconventional superconductivity at low temperatures,^{13,14} are considered of great interest because they present the opportunity to study the competition or the coexistence of the two mechanisms.

- (9) Pourovskii, L. V.; Katsnelson, M. I.; Lichtenstein, A. I. Phys. Rev. B: Condens. Matter Mater. Phys. 2006, 73, 060506.
- (10) Grin, Y. N.; Yarmolyuk, Y. P.; Gladishevsky, E. I. Z. Kristallogr. 1976, 24, 242.
- (11) Moshopoulou, E. G.; Fisk, Z.; Sarrao, J. L.; Thompson, J. D. J. Solid State Chem. 2001, 158, 25.
- (12) Macaluso, R. T.; Sarrao, J. L.; Pagliuso, P. G.; Moreno, N. O.; Goodrich, R. G.; Browne, D. A.; Fronczek, F. R.; Chan, J. Y. J. Solid State Chem. 2002, 166, 245.
- (13) Petrovic, C.; Pagliuso, P. G.; Hundley, M. F.; Movshovich, R.; Sarrao, J. L.; Thompson, J. D.; Fisk, Z.; Monthoux, P. J. Phys.: Condens. Matter 2001, 13, L337.

^{*} To whom correspondence should be addressed. E-mail: cbelin@ univ-montp2.fr (C.B.), mtillard@univ-montp2.fr (M.T.). Fax: 33 04 67 14 33 04. Tel: 33 04 67 14 48 97.

We were especially interested in focusing on the system Sm–Co–Ga and in searching for the compound SmCoGa₅. This work was motivated by the fact that Sm, with an electronic configuration and chemical properties close to those of Pu, is a much more accessible element. A look at the previous literature indicates that this Sm–Co–Ga ternary system is very complex since 18 phases have been reported from the Sm–Co–Ga phase diagram¹⁵ and collected in the Pearson crystal structures handbook.¹⁶

Experimental Section

One incentive for the investigation of ternary phases with Sm and Co stems from the existence of numerous stable compounds: five in Sm–Ga and eight in Sm–Co binary systems whereas only one has been found in the Co–Ga system.¹⁷ Although Ga has a very low melting point (303 K), a property which makes it an excellent medium for flux syntheses, alloying it with elements Sm and Co that melt respectively at 1345 and 1768 K requires high temperatures that are hardly affordable in classical furnaces. In arcmelting, a major problem is the weight loss in elements that have the lowest boiling points; this loss can be more or less reduced by properly adjusting the intensity of the electrode. Although many of our high-temperature furnace and arc-melting experiments have given nonhomogeneous products, we at last succeeded in preparing the hitherto unknown SmCoGa₅, SmCoGa₄, and Sm₄Co₃Ga₁₆ compounds and in determining their crystal structures.

As revealed by the X-ray powder pattern, a typical hightemperature (1373 K) alloying of the elements in the atomic ratio of 1:1:5 inside Ta tubes filled with Ar yielded a mixture of the binary compounds: CoGa₃, Co₅Sm, Co₃Sm, and Ga₃Sm₅. After that, the product was finely ground, pressed into a pellet, and then arcmelted. The resulting roundish ingot exhibited a few tiny single crystals on its surface, and they were characterized by energydispersive X-ray analysis (EDX) as SmCoGa₄ (Sm/Co/Ga ratio of 0.170(2)/0.165(2)/0.661(3); chemical standards are Co, GaAs, and Sm₃Ga₅O₁₂). The structure of the compound was determined from single-crystal X-ray intensity measurements.

The same reaction from the composition ratio of 1:1:5 was monitored in the classical furnace at the higher temperature of 1573 K and was then followed by slow cooling. The resulting product appeared fairly homogeneous and well crystallized; the powder pattern was indexed on the basis of one major component of Sm_4 - Co_3Ga_{16} and small amounts of the side compounds of $SmGa_3$ and $CoGa_3$. The structure of $Sm_4Co_3Ga_{16}$ was also determined from single-crystal X-ray intensities, and it agrees with the chemical EDX analysis (Sm/Co/Ga ratio of 0.176(2)/0.137(2)/0.687(3)).

Regular-shaped crystals of Sm₄Co₃Ga₁₆ were chosen for electrical resistivity measurements using a standard four-probe method in the cryostat of a vibrating sample magnetometer (VSM) from Oxford Instruments, and measurements were performed in the range of 1.6–300 K with zero field under atmospheric pressure. Magnetic properties were measured using a superconducting quantum inter-

- (14) Petrovic, C.; Movshovich, R.; Jaime, M.; Pagliuso, P. G.; Hundley, M. F.; Sarrao, J. L.; Fisk, Z.; Thompson, J. D.; Monthoux, P. *Europhys. Lett.* **2001**, *53*, 354.
- (15) Sichevich, O. M.; Grin, Y. N.; Yarmolyuk, Y. P. Russ. Metall. Engl. Transl. 1989, 2, 198.
- (16) Villars, P.; Calvert, L. D. Pearson's Handbook of Crystallographic Data for Intermetallic Phases, 2nd ed.; ASM International: Materials Park, OH, 1991.

ference device magnetometer MPMS XL7, in a temperature range of 1.8-300 K and in a field range of 0-7 T. The temperaturedependent susceptibility was measured using the ac procedure. The sample was cooled to 1.8 K under zero magnetic field and measured from 1.8 to 300 K (zero-field cooled, ZFC). The field-cooled (FC) measurements were performed with an applied field during the cooling.

It was impossible to prepare SmCoGa₅ as a pure compound by the arc-melting of a mixture of the elements either taken strictly in the appropriate proportions or with an excess of Ga to compensate for its loss by evaporation. The other method used to prepare SmCoGa₅ consisted of alloying the binary compounds of CoGa₃ and SmGa₃. These had been first prepared from the elements in Ta tubes by high-temperature melting (note that in the Sm-Ga binary SmGa₃ represents a specific composition inside the single ϵ -hexagonal phase domain that ranges from SmGa₄ to SmGa₂).^{18,19} Once prepared, the two binary products were finely ground, mixed intimately in a 1:1 proportion, pressed into a pellet, and then arcmelted. The resulting chunk appeared rather homogeneous under a microscope, but after it was broken into small pieces, no singlecrystalline material could be isolated. The X-ray diffraction powder pattern lines were indexed on the basis of a large proportion (\sim 90%) of the compound SmCoGa₅, a small amount of SmCoGa₄, and a monoclinic impurity of unknown composition. The powder pattern was clear and precise enough to allow the Rietveld refinement of SmCoGa₅ in the same space group (*P4/mmm*) as that of PuCoGa₅.

X-ray Analysis. X-ray powder patterns were recorded on a Philips analytical X'pert diffractometer equipped with a Cu tube, a hybrid monochromator (parabolic multilayer mirror and twocrystal monochromator, Cu K $\alpha_{1,2}$ radiation, $\lambda = 1.5405$ and 1.5443 Å), and the Xcelerator detector.

Single crystals were selected under a microscope, placed at the tip of a glass fiber, and then mounted on the Xcalibur CCD (Oxford Diffraction) four-circle diffractometer for intensity measurements. The single-crystal structures were solved and refined with the programs *SHELXS* 97²⁰ and *SHELXL* 97,²¹ and Rietveld analyses were carried out using the program *LHPM-Rietica*.²² Single-crystal refinement data for Sm₄Co₃Ga₁₆ and SmCoGa₄ are collected in Table 1 whereas Tables 2–7 contain the fractional atomic positions, equivalent isotropic thermal displacement parameters, and selected bond distances for SmCoGa₅, Sm₄Co₃Ga₁₆, and SmCoGa₄.

Calculation Methods. DFT calculations of the electronic structures were performed using the gradient-corrected GGA-PW91 exchange and correlation functional²³ at the spin-unrestricted (polarized) level.

The program CASTEP²⁴ uses plane-wave basis sets to treat valence electrons and pseudopotentials to approximate the potential

- (18) Yatsenko, S. P.; Semyannikov, A. A.; Semenov, G. B.; Chuntonov, K. A. J. Less-Common Met. 1979, 64, 185.
- (19) Pelleg, J.; Kimmel, G.; Dayan, D. J. Less-Common Met. 1981, 81, 33.
- (20) Sheldrick, G. M. SHELXS 97, A Program for Crystal Structure Determination; University of Göttingen: Göttingen, Germany, 1997.
- (21) Sheldrick, G. M. *SHELXL 97*, A Program for Refining Crystal Structures; University of Göttingen: Göttingen, Germany, 1997.
- (22) Hunter, B. A.; Howard C. J. LHPM-Rietica: A Computer Program for Rietveld Analysis of X-ray and Neutron Powder Diffraction Patterns, version 1.7.7; ANSBTO, Lucas Heights Research Laboratories: Menai, Australia, 1998.
- (23) Perdew, J. P.; Chevary, J. A.; Vosko, S. H.; Jackson, K. A.; Pederson, M. R.; Singh, D. J.; Fiolhais, C. *Phys. Rev. B: Condens. Matter Mater. Phys.* **1992**, *46*, 6671.
- (24) Payne, M. C.; Teter, M. P.; Allan, D. C.; Arias, T. A.; Joannopoulos, J. D. *Rev. Mod. Phys.* **1992**, *64*, 1045. CASTEP is distributed inside the commercial pack: Materials Studio version 3.1.0; Accelrys, Inc.: San Diego, 2004.

Table 1. Crystallographic Refinement Data for $Sm_4Co_3Ga_{16}$ and $SmCoGa_4$

empirical formula	Sm ₄ Co ₃ Ga ₁₆	SmCoGa ₄
fw	473.43	488.17
cryst syst	tetragonal	orthorhombic
space group	P4/mmm	Cmcm
unit cell dimens (Å)	a = 6.0620(5)	a = 4.1246(6)
	b = 6.0620(5)	b = 15.608(2)
	c = 11.1495(9)	c = 6.4556(9)
$V(Å^3)$	409.72(6)	415.6(1)
Ζ	1	4
density	7.675	7.802
abs coeff (mm ^{-1})	42.768	43.129
F(000)	825	851
cryst size (mm ³)	$0.25 \times 0.14 \times 0.06$	$0.14 \times 0.09 \times 0.08$
θ range for data collection (deg)	3.36-32.32	4.10-32.24
reflns collected	6873	3511
independent reflns	474 [R(int) = 0.0441]	423 [R(int) = 0.0432]
completeness (%)	96.1	95.1
refinement method	full matrix leas	t squares on F^2
data/restraints/params	474/0/28	423/0/24
GOF on F^2	1.328	1.203
final indices $[I > 2\sigma(I)]^a$	0.0262	0.0269
	0.0626	0.0562
indices (all data)	0.0273	0.0352
	0.0630	0.0580
largest diff peak and hole (e•Å ⁻³)	2.017/-2.011	1.891/-1.939

^{*a*} R1 = $\Sigma(|F_{o}| - |F_{c}|)/\Sigma|F_{o}|$, wR2 = $[\Sigma[w(F_{o}^{2} - F_{c}^{2})]/[\Sigma w|F_{o}|^{2})^{2}]^{1/2}$.

field of ion cores. We used the ultrasoft pseudopotentials (USPPs) generated for each element according to the Vanderbilt²⁵ scheme, which allows calculations to be performed at a lower energy cutoff than when norm-conserving pseudopotentials are used. As usual for transition elements, a partial core-corrected USPP was used for Co. Since high-lying shallow-core Sm 5s and 5p orbitals may hybridize with 5d and 4f orbitals, ultrasoft pseudopotentials treat them as valence states. The inner 3d levels of Ga were also considered as valence states. The kinetic cutoff energies were set at 330 or 380 eV (medium or fine quality). We used a density-mixing scheme²⁶ and a Monkhorst-Pack uniform grid²⁷ of automatically generated *k* points; in any case, the spacing between the grid points in reciprocal space was taken at less than 0.05 Å⁻¹.

The program DMOL^{3 28} uses numerical basis sets (DND, double numerical and a polarization d function). For Sm, the 5s²5p⁶4f⁶6s² electrons were treated as valence electrons and core electrons were replaced by a density functional semicore pseudopotential that introduces some degree of relativistic effect. Single effective potentials were also used for Co and Ga atoms.

The ADF-BAND program²⁹ uses numerical atomic orbitals in addition to Slater-type orbitals to get an even better description with only a limited number of basis functions. Double- ζ functions were used for Co and Ga whereas a triple- ζ basis set augmented with two polarization functions was used for Sm. Scalar relativistic effects were taken into account with the zero-order regular approximation method.

Results and Discussion

Structural Comparison of Compounds SmCoGa5 and Sm4C03Ga16. SmC0Ga5. Despite the presence, as impurities in the product, of SmCoGa₄ described below and an incompletely characterized monoclinic phase (a = 7.822 Å, b = 7.051 Å, c = 6.169 Å, $\beta = 99.08^{\circ}$), a three-phase Rietveld refinement of the experimental X-ray powder pattern (Figure 1) was successfully achieved for SmCoGa₅ (of HoCoGa₅ type) in the tetragonal space group P4/mmm (R_p = 1.36%, $R_{wp} = 2.03\%$, $R_{Bragg} = 4.64\%$, GOF = 3.35). The refined unit cell parameters, a = 4.2419(3) Å and c =6.8559(5) Å, are on the order of those reported for the isostructural compounds $RCoGa_5$ (R = Y, Gd-Tm, Lu, Pu).^{1,10,30} The specific internal Ga1 z coordinate converged to 0.303 and is consistent with the value of 0.312 reported for the isostructural PuCoGa₅ compound (a = 4.232 Å and c = 6.786 Å).

In this structure (Figure 2), the Ga1 atoms are disposed at the vertices of a flattened cube of which the face (001) is capped by the Ga2 atom and the face parallel to (110) is capped by a Co atom. The three-dimensional (3D) network is built essentially through short Co1–Ga1 (2.5142 Å) and relatively long Ga1–Ga2 (2.969 Å) bonds. The Sm atom, 8-fold coordinated (2.969 Å) to Ga1 and 4-fold coordinated (2.999 Å) to Ga2, sits at the center of a quasi-regular cuboctahedron of Ga.

The isostructural compound CeCoIn₅ was described on the basis of such cuboctahedra¹¹ that constitute also the cubic 3D CeIn₃ compound (AuCu₃-type structure³¹). The cuboctahedra that contain the f elements are fused to each other by sharing faces to form a two-dimensional (2D) network [CeIn₃] parallel to the *ab* plane while the layer [CoIn₂] forms parallelepipeds sharing edges and fused to the CeIn₃ layer through square faces.

Sm₄Co₃Ga₁₆. The tetragonal structure of Sm₄Co₃Ga₁₆ (Figure 3) was solved using X-ray single-crystal data, in space group P4/mmm, with a = 6.0620(5) Å and c =11.1495(9) Å, and refined to R1 (wR2) = 2.62 (6.26)%, GOF = 1.33. At first sight, there is a strong correlation between the tetragonal P4/mmm structures of SmCoGa₅, Sm₄Co₃Ga₁₆, and the one of Sm₂CoGa₈ represented in Figure 4. Work on the series R_2CoGa_8 (R = Sm, Gd, Tb, Dy, Ho, Er, Tm, Lu, Y) was reported 3 decades ago. The structure of Ho₂CoGa₈ was determined from a single crystal on the basis of 122 reflection intensities collected by the Weissenberg technique (P4/mmm, a = 4.217 Å and c = 10.97 Å, R(F) = 0.139).This structure is now referred to as the Ho₂CoGa₈-type structure to which $\text{Sm}_2\text{CoGa}_8^{10}$ (a = 4.264 Å and c = 11.14Å, with no further structural determination to our knowledge) has been claimed to belong.¹⁰

Sm₄Co₃Ga₁₆, in which the entire crystal structure of SmCoGa₅ can be recognized in the range of *z* values from ca. 0.2 to 0.8, can be described as already done for the RMIn₅

⁽²⁵⁾ Vanderbilt, D. Phys. Rev. 1990, B41, 7892-7895.

⁽²⁶⁾ Kresse, G.; Furthmuller, J. Phys. Rev. 1996, B54, 11169-11186.

⁽²⁷⁾ Monkhorst, H. J.; Pack, J. D. Phys. Rev. 1977, B16, 1748.

^{(28) (}a) Delley, B. J. Chem. Phys. 1990, 92, 508. (b) Delley, B. Int. J. Quantum Chem. 1998, 69, 423. (c) Delley, B. J. Phys. Chem. 1996, 100, 6107. DMOL³ is distributed inside the commercial pack: Materials Studio version 3.1.0; Accelrys, Inc.: San Diego, 2004.

⁽²⁹⁾ te Velde, G.; Baerends, E. J.; Philipsen, P. H. T.; Wiesenekker, G.; Groeneveld, J. A.; Kootstra, F.; de Boeij, P. L.; Snijders J. G. BAND2004.01; SCM, Theoretical Chemistry, Vrije Universiteit: Amsterdam, The Netherlands. http://www.scm.com.

⁽³⁰⁾ Hudis, J.; Hu, R.; Broholm, C. L.; Mitrovic, V. F.; Petrovic, C. J. Magn. Magn. Mater. 2006, 307, 301.

⁽³¹⁾ Knebel, G.; Braithwaite, D.; Canfield, P. C.; Lapertot, G.; Flouquet, J. Phys. Rev. 2001, B65, 24425.



Figure 1. Observed (data points), calculated (solid line), and difference (bottom) room-temperature X-ray diffraction powder patterns (Cu K $\alpha_{1,2}$, $\lambda = 1.5405$ and 1.5443 Å) for an alloy prepared by arc-melting of a SmGa₃/CoGa₃ mixture, resulting preponderantly (90%) in SmCoGa₅.



Figure 2. Representation of the SmCoGa₅ unit cell (a = 4.2419 Å and c = 6.8559 Å, tetragonal space group *P4/mmm*). Ga–Ga contacts have been drawn up to 3 Å.

Table 2. Atomic Positional and Isotropic Displacement Parameters in the Tetragonal (*P4/mmm*) Compound SmCoGa₅

- 4	Wyckoff			_	TT (Å 2)
atom	site	X	у	Z	$U_{\rm iso}$ (A ²)
Sm	1a	0	0	0	0.027(1)
Co1	1b	0	0	¹ / ₂	0.024(2)
Ga1	4i	0	1/2	0.3030(2)	0.044(6)
Ga2	1c	1/2	$1/_{2}$	0	0.072(2)

Table 3. Selected Interatomic Distances (Å) and Atom Coordinations in ${\rm SmCoGa}_5$

Co1	$8 \times \text{Gal}$	2.5142(9)	Ga1	$2 \times \text{Col}$	2.5142(9)
Sm	$8 \times Ga1$	2.969(1)	Ga1	$2 \times \text{Sm}$	2.969(1)
Sm	$4 \times \text{Ga2}$	2.9995(1)	Ga2	$4 \times \text{Sm}$	2.9995(1)
Ga1	Ga1	2.701(2)	Ga1	$2 \times \text{Ga2}$	2.969(1)
Ga1	$4 \times \text{Gal}$	2.9995(1)			

and R_2MIn_8 families (R = rare-earth elements)³² on the basis of the sequential stacking of AuCu₃- and PtHg₂-type fragments. As shown in Figure 4, the difference between R₂-CoGa₈ and Sm₄Co₃Ga₁₆ results from the presence of a Co

(32) Pagliuso, P. G.; Thompson, J. D.; Hundley, M. F.; Sarrao, J. L.; Fisk, Z. Phys. Rev. B: Condens. Matter Mater. Phys. 2001, 63, 054426.



Figure 3. Representation of the Sm₄Co₃Ga₁₆ unit cell (a = 6.0620 Å and c = 11.1495 Å, tetragonal space group *P4/mmm*). Note that atoms have been labeled for comparison with SmCoGa₅.

Table 4. Atomic Coordinates and Equivalent Displacement Parametersin the Tetragonal (P4/mmm) Compound $Sm_4Co_3Ga_{16}$

atom	Wyckoff site	x	у	z	$U_{\rm eq}({\rm \AA}^2)^a$
Sm	4i	0	1/2	0.19458(4)	0.0079(2)
Co1	2e	0	1/2	1/2	0.0075(3)
Co2	1a	0	0	0	0.0109(5)
Ga1	8 <i>r</i>	0.24972(8)	0.24972(8)	0.38553(7)	0.0112(2)
Ga2a	2h	1/2	1/2	0.1953(1)	0.0113(3)
Ga2b	2g	0	0	0.2199(2)	0.0109(3)
Ga3	4j	0.2813(1)	0.2813(1)	0	0.0105(2)

^{*a*} U_{eq} is defined as one-third of the trace of the orthogonalized U_{ij} tensor.

atom inside the [SmGa₃] layer. A superstructure (a = 6.062Å and c = 11.149 Å) is then built from Sm₂CoGa₈ (a = 4.264 Å and c = 11.14 Å) by filling with one Co atom over two squares of Ga within the *ab* plane (Figure 4, middle, top). The arising of the Co–Ga bond (2.412 Å) induces the shortening of the Ga3–Ga3 distance from 3.015 to 2.651 Å. To go back to the comparison with SmCoGa₅, within the plane containing Sm, the Ga2 atom (in SmCoGa₅) has given place to Ga2a and Ga2b in Sm₄Co₃Ga₁₆ (Figure 3), and



Figure 4. Representation of the tetragonal *P4/mmm* structures of Sm₂CoGa₈ (left), SmCoGa₅ (middle, bottom), and Sm₄Co₃Ga₁₆ (right). The difference between Sm₄Co₃Ga₁₆ (a = 6.062 Å and c = 11.149 Å) and Sm₂CoGa₈ (a = 4.264 Å and c = 11.14 Å) arises from the presence of one additional Co atom in the square base of the AuCu₃-type layer. A superstructure is then built (middle, top) by filling one Co atom over two squares of Ga, which induces the shortening of the Ga–Ga distance from 3.015 to 2.651 Å.

Table 5. Interatomic Distances (Å) and Atom Coordinations in $\rm Sm_4Co_3Ga_{16}$

-					
Co1	$8 \times \text{Gal}$	2.4945(4)	Co2	$4 \times \text{Ga3}$	2.412(1)
Sm	$4 \times Ga1$	3.0209(6)	Co2	$2 \times \text{Ga2b}$	2.451(2)
Sm	$2 \times \text{Ga2a}$	3.0310(3)	Ga1	$2 \times Co1$	2.4945(4)
Sm	$2 \times \text{Ga2b}$	3.0441(3)	Ga1	Ga1	2.553(2)
Sm	$4 \times \text{Ga3}$	3.0614(4)	Ga1	Ga2b	2.828(1)
Ga2a	$4 \times \text{Ga3}$	2.874(1)	Ga1	$2 \times \text{Sm}$	3.0209(6)
Ga2a	$4 \times \text{Sm}$	3.031(5)	Ga3	Co2	2.412(1)
Ga2b	Co2	2.452(2)	Ga3	$2 \times \text{Ga3}$	2.651(2)
Ga2b	$4 \times Ga1$	2.828(1)	Ga3	$2 \times \text{Ga2a}$	2.874(1)
Ga2b	$4 \times \text{Sm}$	3.0441(3)	Ga3	$4 \times \text{Sm}$	3.0614(4)

subsequently, the unique long distance of 2.969 Å between Ga1 and Ga2 (SmCoGa₅) is split into 3.017 Å (Ga1–Ga2a) and 2.828 Å (Ga1–Ga2b).

The comparative analysis of the interatomic distances is informative on the nature and localization of bonding within these compounds. In all structures, Sm is surrounded by 12 Ga atoms: eight at 2.969 Å and four at 2.999 Å in SmCoGa₅ and eight at 3.007 Å and four at 3.020 Å in Sm₂CoGa₈ instead of four at 3.021 Å, two at 3.031 Å, two at 3.044 Å, and four at 3.061 Å in Sm₄Co₃Ga₁₆.

The shortest interatomic distances involve the eightcoordinated Co atom, and the Co–Ga length is 2.514 Å in SmCoGa₅, 2.510 Å in Sm₂CoGa₈, and 2.495 Å in Sm₄Co₃-Ga₁₆. In the latter, the additional Co atom (Co2) is coordinated to six Ga atoms: four Ga3 at 2.412 Å and two Ga2b at 2.451 Å. In these structures, there are only a few short contacts between Ga atoms along the $\langle 001 \rangle$ direction, and the Ga1–Ga1 length of 2.701 Å in SmCoGa₅ is shortened to 2.649 Å in Sm₂CoGa₈ and to 2.553 Å in Sm₄Co₃Ga₁₆.

With regard to Sm₂CoGa₈, this superstructure gives some good small statistical intensity reflections, having h + k with an odd value, and its simulated powder pattern (Cu K α) represented in Figure 5 is indexed in the two *P*4/*mmm* cells (a = 4.264 Å and c = 11.14 Å; a = 6.062 Å and c = 11.149 Å).

Crystal Structure of SmCoGa₄. The crystal structure of SmCoGa₄ was solved in the orthorhombic space group *Cmcm*, with a = 4.1246(6) Å, b = 15.608(2) Å, and c = 6.4556(9) Å, and refined from X-ray single-crystal intensities to R1 (wR2) = 2.69 (5.62)%, GOF = 1.20. This compound belongs to the YNiAl₄ type and is isostructural with SmNiGa₄³³ and YbNiGa₄.³⁴

As shown in Figure 6, the 3D network of SmCoGa₄ may be described with alternate slabs: a nearly square 2D network of Ga2 and Ga3, puckered by folding along the Ga2–Ga2 square diagonal, and a corrugated 2D slab of Ga1 rings (boat conformation) filled with Co atoms. These slabs, stacked along the *b* axis, are interconnected through very short Co– Ga2 bonds (2.308 Å). The Sm atom lies within a 17-atom cavity (4 Co and 13 Ga atoms). These cavities are fused to each other through the sharing of 5-atom rings, forming channels parallel to the *a* axis.

Electronic Structure Calculations and Properties. As indicated by the electrical resistivity and magnetic measurements on Sm₄Co₃Ga₁₆, all these ternary materials would be expected to exhibit some metallic properties and magnetic moments. According to the Pauling electronegativities ($\chi =$ 1.2, 1.6, and 1.8 for Sm, Ga, and Co, respectively), we might reasonably assign the formal charge 3+ to Sm whereas Co should be the most reduced element in the alloys. To gain

⁽³³⁾ Romaka, V. A.; Grin, Yu.; Yarmoluk, Ya.P. Ukr. Fiz. Zh. 1983, 28 (7), 1095.

⁽³⁴⁾ Vasylechko, L.; Schnelle, W.; Schmidt, M.; Burkhardt, U.; Borrmann, H.; Schwarz, U.; Grin, Yu. J. Alloys Compd. 2006, 416, 35.



Figure 5. Representation of the simulated X-ray powder pattern calculated for $\text{Sm}_4\text{Co}_3\text{Ga}_{16}$ (Cu K $\alpha_{1,2}$). The lines are indexed in the tetragonal *P4/mmm* unit cells: a = 6.062 Å and c = 11.149 Å for $\text{Sm}_4\text{Co}_3\text{Ga}_{16}$ (upper ticks) and a = 4.264 Å and c = 11.14 Å for Sm_2CoGa_8 (lower ticks).



Figure 6. Packing within the orthorhombic structure of SmCoGa₄ (*Cmcm*, a = 4.1246 Å, b = 15.608 Å, and c = 6.4556 Å) approximately viewed along the *a* axis. For clarity, only the atomic contacts less than 2.90 Å are drawn.

Table 6. Atomic Coordinates and Equivalent Displacement Parametersin $SmCoGa_4$ (Orthorhombic *Cmcm*)

	Wyckoff				
atom	site	x	У	z	$U_{\rm eq}({\rm \AA}^2)$
Sm	4c	0	0.88343(4)	1/4	0.0085(2)
Co	4c	0	0.2242(1)	1/4	0.0140(4)
Ga1	8f	0	0.31164(6)	-0.0512(1)	0.0107(2)
Ga2	4c	0	0.07626(9)	1/4	0.0135(3)
Ga3	4a	$1/_{2}$	0	0	0.0128(3)

some insight into the understanding of these phases and bonding therein, we have calculated the electronic structures for the three compounds.

First-principles calculations relative to the lanthanide and actinide compounds have been widely discussed. In the systems containing f elements, strong electron correlations cannot be correctly described in the framework of the conventional DFT-LDA (LDA = local density approximation) theory. Such treatment may lead to erroneous electronic properties, for example, as in the case of UO₂, which was

Table 7.	Interatomic	Distances	(Å)	and	Atom	Coordinations	iı
SmCoGa ₄							

Ga2	3.010(1)	Sm	$4 \times \text{Gal}$	3.0480(7)
$4 \times \text{Ga3}$	3.1888(4)	Sm	$2 \times Co$	3.230(1)
Ga2	3.2889(6)	Co	Ga2	2.308(2)
$4 \times Ga1$	2.4923(7)	Co	$2 \times \text{Sm}$	3.230(1)
Co	2.376(1)	Ga1	$2 \times Co$	2.4923(7)
Ga1	2.566(2)	Ga1	$2 \times Ga1$	2.897(1)
2 x Sm	3.0480(7)	Ga1	Sm	3.304(1)
Co	2.308(2)	Ga2	$4 \times \text{Ga3}$	2.8765(6)
Sm	3.010(1)	Ga2	$2 \times \text{Sm}$	3.2886(5)
$4 \times \text{Ga2}$	2.8765(6)	Ga3	$4 \times \text{Sm}$	3.1888(4)
	$\begin{array}{c} Ga2\\ 4\times Ga3\\ Ga2\\ 4\times Ga1\\ Co\\ Ga1\\ 2\ x\ Sm\\ Co\\ Sm\\ 4\times Ga2\\ \end{array}$	$\begin{array}{cccc} Ga2 & 3.010(1) \\ 4 \times Ga3 & 3.1888(4) \\ Ga2 & 3.2889(6) \\ 4 \times Ga1 & 2.4923(7) \\ Co & 2.376(1) \\ Ga1 & 2.566(2) \\ 2 x Sm & 3.0480(7) \\ Co & 2.308(2) \\ Sm & 3.010(1) \\ 4 \times Ga2 & 2.8765(6) \end{array}$	$\begin{array}{cccccc} Ga2 & 3.010(1) & Sm \\ 4\times Ga3 & 3.1888(4) & Sm \\ Ga2 & 3.2889(6) & Co \\ 4\times Ga1 & 2.4923(7) & Co \\ Co & 2.376(1) & Ga1 \\ Ga1 & 2.566(2) & Ga1 \\ 2 x Sm & 3.0480(7) & Ga1 \\ Co & 2.308(2) & Ga2 \\ Sm & 3.010(1) & Ga2 \\ 4\times Ga2 & 2.8765(6) & Ga3 \\ \end{array}$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$

found to be metallic instead of insulating. However, good results may be obtained using GGA gradient-corrected functionals in cases where spin—orbit interactions have little influence on the structural properties.³⁵ To calculate accurate structural energies,³⁶ relativistic effects have to be included in the calculations.

SmCoGa₅. Among the three compounds studied in this work, SmCoGa₅, with only seven atoms in its unit cell, is the most computationally affordable at the ab initio level.

A geometry optimization was first performed with CASTEP (P4/mmm symmetry-constrained, BFGS algorithm) by simultaneously varying the cell parameters and the *z* internal coordinate of the Ga1 atom. Results from spin-unrestricted and spin-restricted calculations are compared in Table 8.

Calculated lattice constants deviate from the experimental values by less than 1.8%, and the *z* coordinate of Ga1 deviates by less than 3.8%. This looks good with regard to the structural parameters obtained from a multiphase powder refinement. However, the lattice parameters of SmCoGa₅ from the unrestricted calculation are in poorer agreement with the experiment than those from the spin-restricted one. This

Jia et al.

⁽³⁵⁾ Pickard, C.; Winkler, B.; Chen, R. K.; Payne, M. C.; Lee, M. H.; Lin, J. S.; White, J. A.; Milman, V.; Vanderbilt, D. *Phys. Rev. Lett.* 2000, 85 (24), 5122.

⁽³⁶⁾ Soderlind, P. Adv. Phys. 1998, 47, 959.





Figure 7. Representation of partially polarized DOS for compound SmCoGa₅ calculated with CASTEP (top) and DMOL³ (bottom). In the nonrelativistic calculation (CASTEP), the Sm 4f levels are pinned at the Fermi level with only spin-up density (spin charge 6.12). Scalar relativistic effects are introduced in the DMOL³ calculation (spin charge 5.806).

Table 8. Comparison of Experimentally Determined (Room Temperature) and CASTEP DFT Optimized Parameters of SmCoGa₅ in Spin-Restricted and Spin-Unrestricted Configurations

		experimental	spin-restricted	unrestricted
a (Å)		4.2419	4.2648 (+0.5%)	4.2888 (+1.1%)
<i>c</i> (Å)		6.8559	6.8921 (+0.5%)	6.9799 (+1.8%)
Ga1(z)		0.3030	0.3102 (+2.3%)	0.3145 (+3.8%)
total energy (eV)			-13514.49	-13530.11
spin (ħ)	Sm			3.13
	Co			-0.02
	Ga1/Ga2			0.01/-0.01
charge	Sm		0.09	0.25
	Co		-0.56	-0.59
	Ga1/Ga2		-0.06/+0.13	0.03/0.08

might stem from the failure of the local spin density approximation, generalized gradient approximation (LSDA-GGA) to treat correctly the strongly correlated systems with localized 4f states (narrow bandwidth) at $E_{\rm F}$. However, this discrepancy due to the strong on-site Coulomb repulsion among the 4f electrons might be corrected with the DFT+U method in which a Hubbard-like term is added to the DFT Hamiltonian.³⁷

The calculated electronic densities of states (DOS) indicate that SmCoGa₅ is likely to display some metallic character (Figure 7), and the total energy has been found to be 2.23 eV/atom lower for the unrestricted (spin-polarized) configuration.

Table 9. Mulliken Atomic and Spin Charges Calculated with CASTEP, DMOL³, and ADF-BAND for the Compound SmCoGa₅

	CAS	CASTEP		DMOL ³		ADF-BAND	
	atomic charge	spin charge	atomic charge	spin charge	atomic charge	spin charge	
Sm Ga1 Ga2a Co1	$\begin{array}{r} 0.340 \\ 0.050 \\ -0.140 \\ -0.390 \end{array}$	$6.200 \\ -0.020 \\ -0.040 \\ -0.140$	2.506 -0.295 -0.649 -0.675	5.806 0.009 0.001 -0.090	$\begin{array}{r} 2.358 \\ -0.338 \\ -0.555 \\ -0.451 \end{array}$	5.545 0.006 0.003 0.009	

The Mulliken atomic and spin charges obtained with the CASTEP, DMOL3, and ADF-BAND calculations are reported for SmCoGa₅ in Table 9. Note that the CASTEP atomic charges are fairly different from those obtained with DMOL³ and ADF-BAND. In fact, the Mulliken charges and bond populations are calculated in CASTEP following the formalism described by Segall et al.^{38,39} Due to the delocalized nature of the basis states, plane waves do not give information regarding the localization of the electrons in the system; in contrast, the use of linear combination of atomic orbitals (LCAO) basis sets provides a natural way of specifying atomic charges and bond populations. In CASTEP, population analysis is performed using a projection of the plane-wave states onto a localized basis,40 to which the Mulliken formalism⁴¹ is then applied. On the other hand, DMOL³ and ADF-BAND, which incorporate relativistic effects, together yield very comparable band structures, DOS, spins, and atomic charges. As a consequence, the Sm 4f orbitals which are found pinned at the Fermi level in the nonrelativistic CASTEP calculation are split on both sides of $E_{\rm F}$ when using ADF-BAND and DMOL³ (Figure 7).

Sm₄Co₃Ga₁₆. The band structure and DOS are comparatively represented for the two compounds Sm₄Co₃Ga₁₆ and SmCoGa₅ (Figure 8). These electronic structures are clearly different, emphasizing the more 2D nature in conduction properties for SmCoGa5 (weak dispersion with no band crossing the Fermi level along the [001] direction). The electronic properties look more isotropic for Sm₄Co₃Ga₁₆ which, with a higher DOS at the Fermi level, appears more metallic than SmCoGa₅. Actually, as indicated by the Mulliken atomic charges given in Table 10, in Sm₄Co₃Ga₁₆ the positive charge on Sm is less than that in SmCoGa₅ and the anionic charge is more evenly distributed over the Ga atoms. For both compounds, within the energy range of ~ 3 eV (~0.1 hartree) below $E_{\rm F}$, the DOS are mainly governed by overlapping Co 3d and Ga 4p orbitals whereas the little dispersed Sm 4f orbital remains practically inert.

The electron-density differences calculated with CASTEP for SmCoGa₅ and Sm₄Co₃Ga₁₆ provide a good overview of the electronic charge redistribution due to some interatomic bonding in the system (Figure 9). In SmCoGa₅, 2D-like bonding within a layer perpendicular to the c axis involves the atoms Co1 and Ga1. There is no apparent electron

⁽³⁷⁾ Petersen, M.; Hafner, J.; Marsman, M. J. Phys.: Condens. Matter 2006, 18, 7021.

⁽³⁸⁾ Segall, M. D.; Pickard, C. J.; Shah, R.; Payne, M. C. *Mol. Phys.* **1996**, *89*, 571.

 ⁽³⁹⁾ Segall, M. D.; Shah, R.; Pickard, C. J.; Payne, M. C. *Phys. Rev.* 1996, *B54*, 16317.

⁽⁴⁰⁾ Sanchez-Portal, D.; Artacho, E.; Soler, J. M. Solid State Commun. 1995, 95, 685.
(41) Mulliken, R. S. J. Chem. Phys. 1955, 23, 1833.



Figure 8. Representation of band structures and total DOS calculated with $DMOL^3$ for compounds $SmCoGa_5$ and $Sm_4Co_3Ga_{16}$.

Table 10. Compared Mulliken Atomic and Spin Charges (DMOL 3 and GGA-PW91)

	SmCoGa ₅		Sm ₄ C	o_3Ga_{16}	$SmCoGa_4$	
atom	atomic charge	spin charge	atomic charge	spin charge	atomic charge	spin charge
Sm	2.506	5.806	1.970	5.810	2.678	5.830
Ga1	-0.295	0.009	-0.216	-0.008	-0.442	0.013
Ga2a	-0.649	0.001	-0.489	0.010	-0.518	0.005
Co1	-0.675	-0.090	-0.678	-0.070	0.176	1.437
Ga2b			-0.518	0.005		
Ga3			-0.492	0.015	-0.549	-0.009
Co2			-0.816	0.065		

density that bounds Ga2, the more reduced Ga atom, to neighboring Ga atoms. Also, at the isovalue level of 0.032 $e^{\hat{A}^{-3}}$, no density is visible around Sm, featuring its cationic nature. For compound Sm₄Co₃Ga₁₆, bonding at the middle layer (Co1–Ga1) looks like that in SmCoGa₅; in addition, there is consistent bonding within the basal layers (Co2– Ga3 and Ga3–Ga3). On the other hand, there is some weak bonding between these layers involving the Co2, Ga2b, and Ga1 atoms that provides the structure with some degree of three-dimensionality.

Electrical and Magnetic Properties of Sm₄Co₃Ga₁₆. Sm₄-Co₃Ga₁₆ is the only compound that could be prepared pure, and the results from the electrical and magnetic measurements follow. Sm₄Co₃Ga₁₆ presents a resistivity of 4.2 $\mu\Omega$ · cm at 3 K. The resistivity reaches the value of 25.0 $\mu\Omega$ ·cm at 300 K with a linear behavior as a function of temperature in the temperature range from 3 to 300 K, which is classical for a rather good metal. The compound is found to be paramagnetic above 3.5 K, with an effective moment of 1.36 $\mu_{\rm B}$ /unit (or 0.34 $\mu_{\rm B}$ considering one Sm atom) and a Weiss



Figure 9. Representation of the electron-density difference (isovalue level $0.032 \text{ e}^{-}\text{Å}^{-3}$) in the SmCoGa₅ (left) and Sm₄Co₃Ga₁₆ (right) compounds. The electron-density difference is obtained by subtracting the isolated atom densities from the overall electron density.



Figure 10. Electrical resistivity measurements for compound Sm_4Co_3 -Ga₁₆ using a four-probe method. The results are shown only for temperatures below 6 K.

temperature of -17 K, denoting a slight antiferromagnetic ordering. The low value of the experimental magnetic moment compared with the computed value (close to $6 \mu_{\rm B}/$ Sm) is expected from the large cancellation of the spin moment by the orbital contribution, as reported for PuCoGa₅.⁴² As pointed out in a previous study of the ternary phase Sm₂Co_{17-x}Ga_x (x = 0-7),⁴³ the Ga substitution for Co leads to an approximately linear decrease in the Curie temperature and a rapid decrease in the saturation magnetic moment, which is faster than that in the case of magnetic dilution. Therefore, the absence of ferromagnetism in the compounds studied, which are rich in Ga, is not surprising.

From magnetic and electrical measurements, we show the evidence for bulk superconductivity in $Sm_4Co_3Ga_{16}$. A transition to zero resistance is observed near 2.8 K (Figure 10); this value is almost co-incident with the shielding diamagnetism observed at 3.2 K (Figure 11). The full-shielding diamagnetism is not observed, certainly in relation with the presence of paramagnetic impurities in the bulk sample. The discrepancy between these two values is due to a slight offset of the temperature probe in the VSM cryostat. This temperature is very close to the value of 2.3 K found

⁽⁴²⁾ Opahle, I.; Oppeneer, P. M. Phys. Rev. Lett. 2003, 90, 157001.

⁽⁴³⁾ Liang, B.; Kleinschroth, I.; Shen, B.; Kronmuller, G., H. J. Appl. Phys. 2000, 87, 5314.



Figure 11. ZFC/FC magnetic susceptibility measurements for compound $Sm_4Co_3Ga_{16}$ (ac mode).



Figure 12. Representation of the electron-density difference (isovalue level $0.021~e^{*} {\rm \AA}^{-3})$ calculated for SmCoGa4.

for the HF CeCoIn₅ compound, which displays a resistivity of 8 $\mu\Omega$ ·cm at 3 K. The studies on the superconductivity of PuCoGa₅ have suggested the occurrence of unconventional superconductivity. In such a scenario, the order-of-magnitudelower T_c value for Sm₄Co₃Ga₁₆ as compared with that of PuCoGa₅ ($T_c = 18.5$ K) would result from a decrease in the bandwidth in going from 5f to 4f electrons that have a greater degree of localization. In the case of Sm₄Co₃Ga₁₆, superconductivity should almost exclusively be driven by latticevibration pairing.

SmCoGa₄. The assumption that in these compounds Sm bears a 3+ formal charge in agreement with its lowest electronegativity is supported not only by the DFT calculation but also by the magnetic measurements carried out for Sm₄-Co₃Ga₁₆. This is very well illustrated in the electron-density difference map (Figure 12) calculated for SmCoGa₄ in which the Sm atom bears a 2.678+ charge and in which the main bonding involves the Co and Ga1 atoms (see Figure 6 for atom labeling). With regard to the Mulliken atomic charges (Table 10), the structure can be roughly divided into two anionic domains, the corrugated Co/Ga1 layer with a mean atomic charge of 0.236- and the 2D square network (mean atomic charge of 0.533-) in which the bonding is weaker

(Ga2-Ga3: 2.8765 Å). The electron-density difference map also gives evidence of the strong polarization that the Sm cation exerts on bonds forming the 17-vertex cage in which it is lying.

Concluding Remarks

In this paper, we have reported the existence of three novel compounds in the ternary system Sm-Co-Ga. SmCoGa₅ (I) displays the same tetragonal structure as PuCoGa₅ and Sm₄Co₃Ga₁₆ (**II**) is strongly related to the SmCoGa₅ structure, although the orthorhombic compound SmCoGa₄ (III), of the YNiAl₄ type, is quite different. The common feature of these compounds is that they are built from the stacking of atomic layers: plane SmGa and crown-CoGa4 for I and plane CoGa₄, quasi-plane SmGa, and crown-CoGa₄ for **II**. The structure of III is composed of corrugated layers of Ga, layers of Sm, and corrugated layers of CoGa2. SmCoGa5 and PuCoGa₅ have the same tetragonal structure as the quasi-2D HF CeCoIn₅ compound that displays a superconducting transition at 2.3 K. The properties of the latter have been discussed in consideration of the stacking along the 4-fold axis of monolayers of the face-sharing distorted cuboctahedra CeIn₃ and the monolayers of edge-sharing rectangular parallelepipeds CoIn2.44 In addition to their electronic descriptions, the origin and level of superconduction in these systems may be correlated with their geometrical features, as for the CeMIn₅ series.⁴⁴

With the most distorted cuboctahedra, CeCoIn₅ and CeIrIn₅ exhibit ambient pressure superconductivity below 2.3 and 0.4 K, respectively, whereas CeRhIn₅, in which the CeIn₃ cuboctahedron is the most regular, is antiferromagnetic at ambient pressure with $T_{\rm N} = 3.8$ K, like the cubic CeIn₃ ($T_{\rm N}$ = 10 K). Interestingly, superconductivity can be induced in the latter at a very low T_c (below 200 mK) by applying a pressure of 2.5 GPa. Now, the comparison with PuCoGa₅ is worthy because in this compound Pu is located at the center of a very regular cuboctahedron (quasi-perfect $m\bar{3}m$ symmetry). The superconductivity in PuCoGa₅, which experiences a T_c of 18.5 K (22 K at 16.4 GPa),² arises from a pairing mechanism either mediated by lattice vibrations with electron pairs having no relative angular momentum (isotropic s-wave) or magnetically mediated (p or d wave), as is the case for HF materials and high T_c superconductors.

In SmCoGa₅, the distortion of the SmGa₃ cuboctahedron is roughly similar to that found for the CeIn₃ cuboctahedra in CeCoIn₅ and CeIrIn₅, suggesting possible superconducting properties that have not been investigated because SmCoGa₅ could not be obtained as a pure phase.

However, the study of these new ternary compounds seems promising if one considers the evidence for superconductivity in the $Sm_4Co_3Ga_{16}$ compound. Indeed, the close relationship between the distortion in the local environment of the f elements (Pu in the case of PuCoGa₅) and the value of the critical temperature observed for the actinide-based com-

⁽⁴⁴⁾ Moshopoulou, E. G.; Sarrao, J. L.; Pagliuso, P. G.; Moreno, N. O.; Thompson, J. D.; Fisk, Z.; Ibberson, R. M. Appl. Phys. A 2002, 74 (suppl. 1), S895.

pounds was qualitatively corroborated by our results. The T_c measured on our Sm₄Co₃Ga₁₆ sample (~3 K), albeit low compared with any technological requirement, could be varied by considering other parameters like stoichiometry and pressure. Furthermore, this family is based on nonradioactive elements, which allow a comprehensive study

without protracted precautions and without any temporal evolution of the compounds.

Supporting Information Available: One CIF file. This material is available free of charge via the Internet at http://pubs.acs.org. IC0700755